Roll No.:

Total No. of Questions: 9]

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91034

B.Sc. (Chemistry Hons.) 1st Semester (Full & Reappear)
Examination, March-2021
(w.e.f. 2012-13)

PHYSICAL CHEMISTRY

Time: Three Hours]

[Maximum Marks : 40

Before answering the questions, candidates should ensure that they have been supplied the correct and complete question paper. No complaint in this regard, will be entertained after examination.

- Note: Attempt five questions in all, selecting one question from each Section. Question No. 1 is compulsory.

 All questions carry equal marks.
- (a) Calculate the various degree of freedom for CO₂ molecule.

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- (b) What is Compressibility Factor?
- (c) Define root mean square velocity.
- (d) Give one example each for absorption and adsorption.
- (e) What is Space Lattice?
- (f) Name the instrument used for measuring interfacial angles in crystal.
- (g) Define Coefficient of Viscosity.
- (h) How does evaporation helps in cooling?

 1×8=8

Section-A

- 2. (a) Describe the following:
 - (i) Graham's law of diffusion
 - (ii) Critical constants
 - (b) Derive van der Waals equation of state for real gases.
- 3. (a) State and explain principle of equipartition of energy.
 - (b) Derive reduced equation of state.
 - (c) Write a note on liquification of gases. 4,2,2

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Section-B

- 4. (a) Explain the following:
 - (i) Viscosity of gases
 - (ii) Collision frequency
 - (b) What is mean free path? Derive an expression for it.
- 5. (a) Explain the following:
 - (i) Adsorption isobaco
 - (ii) Physical adsorption
 - -(iii) Chemical adsorption
 - (b) Explain and compare Langmuir and Freundlieh adsorption isotherms. 3,5

Section-C

- (a) Differentiate between Crystalline and Amorphous solids.
 - (b) Explain the following:
 - (i) Axis of symmetry
 - (ii) Mechanism of enzyme catalysis
 - (iii) Law of rational indices

2,6

- 7. (a) Describe Bravais lattices.
 - (b) Derive Gibbs adsorption equation.

Section-D

- 8. (a) What is surface tension and surface energy? How surface tension varies with temperature?
 - (b) Explain liquid crystal and its types. 4,4
- 9. Describe the following:
 - (i) Rheochor
 - (ii) Parachor
 - (iii) Thermography
 - (iv) Vapour pressure

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